

Course Code: 23BS2T05  
BONAM VENKATA CHALAMAYYA INSTITUTE OF TECHNOLOGY &  
SCIENCE  
(AUTONOMOUS)

*I - B. Tech II Semester Regular Examinations (BR23), June - 2025*

**ENGINEERING CHEMISTRY (CE)**

Time: 3 hours

Max. Marks: 70

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*Question Paper consists of Part-A and Part-B*  
*Answer ALL the question in Part-A and Part-B*

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PART-A (10X2 = 20M)

	Marks	CO	BL
1. a) Define hard water.	(2M)	CO1	L1
b) What is meant by desalination	(2M)	CO1	L1
c) What is a secondary battery?	(2M)	CO2	L2
d) Define cathodic protection.	(2M)	CO2	L2
e) What is polymerisation?	(2M)	CO3	L1
f) Define HCV.	(2M)	CO3	L1
g) Define a lubricant.	(2M)	CO4	L1
h) What is a flash point?	(2M)	CO4	L1
i) What is a colloid?	(2M)	CO5	L1
j) Define Viscosity index.	(2M)	CO5	L2

PART-B (5X10 = 50M)

2.a) Discuss about the causes, effects and prevention of Scale and Sludge? (10M) CO1 L2

(OR)

3 a) List the specifications of drinking water (5M) CO1 L1  
b) Describe the desalination of brakish water by Reverse osmosis (5M) CO1 L2

4.a) Interpret Nernst equation in determine the potential of a single electrode. (5M) CO2 L3

b) Discuss about the factors affecting corrosion. (5M) CO2 L2

(OR)

5 a) Discuss dry corrosion with pilling Bedworth rule (5M) CO2 L2

b) Describe the working of Zn-air cell. (5M) CO2 L2

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|------|--|-------|-----|----|
| 6 a) | Differentiate between Thermoplastic and Thermosetting plastics.                          | 5(M)  | CO3 | L4 |
| b)   | Write a note on preparation, properties and uses of nylon-6,6.                           | 5(M)  | CO3 | L1 |
| (OR) |  |       |     |    |
| 7 a) | Interpret on proximate analysis of coal  | 5(M)  | CO3 | L3 |
| b)   | Explain about the refining of petroleum.   | 5(M)  | CO3 | L2 |
| 8.a) | Discuss about fibre rein forced composites explaining their properties and applications. | 10(M) | CO4 | L2 |
| (OR) |  |       |     |    |
| 9 a) | Discuss on the classification of composites  | 5 (M) | CO4 | L2 |
| b)   | List out the applications of refractories.   | 5(M)  | CO4 | L1 |
| 10   | Interpret Braggs method in synthesis of colloids.  | 5(M)  | CO5 | L3 |
| a)   |  | 5(M)  | CO5 | L2 |
| b)   | Explain the formation of micelle in colloids.  |       |     |    |
| (OR) |  |       |     |    |
| 11   | Discuss about adsorption isotherm  | 5(M)  | CO5 | L2 |
| a)   |  |       |     |    |
| b)   | Explain chemical methods of preparing Nano metals.                                       | 5(M)  | CO5 | L2 |

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